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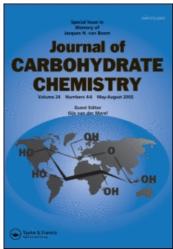
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SYNTHESIS OF BIS(CARBAMOYL ESTER) DERIVATIVES OF D-GLUCOSE AS ANTIFUNGAL PRODUCTS

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ABSTRACT

Regiospecific carbamoylation of di-O-isopropylidene protected D-glucose gave a series of carbamoyl, thiocarbamoyl and dithiocarbamoyl esters at the C-3 secondary site. These were partially and fully deprotected to give two series of derivatives such that a choice of the extent of hydrophilic character could be made. The partially protected products were further derivatized regioselectively with a second carbamoyl, thiocarbamoyl or dithiocarbamoyl group at the C-6 primary site. We also report on antifungal properties of some of those compounds.

INTRODUCTION

Carbamates are important in various areas of medicinal chemistry. 1-4 Also N-alkylcarbamoyl and N,N-dialkylcarbamoyl esters are in current use as pesticides insecticides, 6.7 fungicides, 8.9 and herbicides. 10 These series include dithiocarbamic acid salts as well as carbamoyl, thiocarbamoyl and dithiocarbamoyl esters. Their biological properties vary in accordance with the type of carbamoyl group 11 (thiocarbamate and dithiocarbamate) such as when either or both oxygen atoms are replaced by sulphur atoms and when the nitrogen atom has one or two alkyl groups. We previously demonstrated 12

that compounds with multiple substitutions of carbamoyl groups on a low molecular weight polyol such as glycerol (1,3-dideoxy-bis-1,3-S-(N,N-diethyldithiocarbamoyl)-glycerol) showed enhanced antifungal activity whilst having low toxicity (in contrast to many dithiocarbamic acid salts which release trace toxic metals¹³). This encouraging result led us to explore the synthesis of bis(carbamoyl esters) of carbohydrates with the objective of obtaining less hazardous and more environmentally friendly compounds. D-Glucose was chosen as the carbohydrate moiety since it is cheap and can be readily protected to permit:
(i) a one-step bis-carbamoylation; (ii) both regiospecific carbamoylation at the secondary C-3 position and regioselective esterification at the primary C-6 position; (iii) a modulation of the hydrophilic-lipophilic balance (HLB) by choice of the number of free hydroxyl groups remaining in the final product; (iv) a convenient choice of either furanose or pyranose derivatives. This approach conveniently allows the preparation of a large range of biscarbamates in which the two carbamoyl groups may be the same or different. Antifungal activities of some derivatives are reported.

RESULTS AND DISCUSSION

The range of bis(carbamoyl esters) reported in this paper are synthesized from 1,2:5,6-di-O-isopropylidene- α -D-glucofuranose (1). This starting material has provided convenient access to such products using two strategies. In the first strategy, the diesterification is effected in a one-step reaction following deprotection of the 5,6-isopropylidene group and consequent activation at both the 3- and 6- positions on 1,2-O-isopropylidene- α -D-glucofuranose. In the second strategy, the first ester group is attached at the C-3 position of the diacetoneglucose and the second, identical or not, is attached regiospecifically or regioselectively at the C-6 position, after 5,6- acetal deprotection.

Synthesis of bis(dithiocarbamoyl esters) by one-step biscarbamoylation. (Scheme 1) The key intermediate in the synthesis of the bisdithiocarbamates is the 3-iodo-6-O-tosyl-derivative 1'. This was prepared by first converting compound 1 into the corresponding 3-iodo-derivative, 14 deprotecting the 5,6- acetal group and finally regionselectively tosylating the 6-OH group. The diactivated compound 1' was reacted

a) I₂, PPh₃, imidazole, toluene, 110 °C, 180 min, 69%; b) 0.6M HCl, water, dioxane, 30 °C, 30 min, 64%; c) *p*-toluenesulphonyl chloride, pyridine, toluene, 4 °C, 48 h, 84%; d) (C₂H₅) ₂NCSSLi, HMPA, toluene, 110 °C, 24 h, 60%.

Scheme 1

with excess lithium N,N-diethyl dithiocarbamate¹⁵ in HMPA-toluene (1:1) at 110 °C to give the bis(N,N-diethyldithiocarbamoyl ester) 2 (24 h, 60%). NMR data are in Tables 1-3.

General synthesis of bis(carbamoyl esters) by successive carbamoylations. Biscarbamate derivatives with two identical or different carbamoyl groups were synthesized in either three or four steps from compound 1 (Scheme 2). The hydrophilic character can be increased by deprotection of the remaining 1,2- acetal group in the final step of the synthesis. NMR data of obtained products are in Tables 1-4.

Step a: The 3-O-(N,N-diethylcarbamoyl) ester 3 was obtained by condensing compound 1 with N,N-diethylcarbamoyl chloride (1.2 equiv) in the presence of finely powdered KOH (2 equiv) in Me₂SO-toluene (5:95) at 8 °C (15 min in 81% yield). Using the same conditions, the rate of the reaction leading to the 3-O-(N,N-diethylthiocarbamoyl) ester 4 from the corresponding diacetonide 1 and N,N-diethylthiocarbamoyl chloride was slow, hence potassium t-butoxide was used instead of KOH to obtain product 4 with an improved yield (85%) and in a shorter reaction time (15 min vs 5 h).

The 3-S-(N,N-diethyldithiocarbamoyl) ester 5 was obtained, in 87% yield, by first converting compound 1 into 3-deoxy-3-iodo-1,2:5,6-di-O-isopropylidene- α -D-allofuranose followed by (S_N2) reaction with lithium dithiocarbamate (1.5 equiv) using similar conditions used for the preparation of the diester 2.

a
$$R_1R_2N$$
 R_1R_2N
 R_1R_2N

2; 17 :
$$X = X' = Y = Y' = S$$
; $R_1 = R_2 = R_3 = R_4 = CH_2CH_3$
3; 6 : $X = Y = O$; $R_1 = R_2 = CH_2CH_3$
4; 7 : $X = O$; $Y = S$; $R_1 = R_2 = CH_2CH_3$
5; 8 : $X = Y = S$; $R_1 = R_2 = CH_2CH_3$
9; 18 : $X = X' = Y = Y' = O$; $R_1 = R_2 = R_3 = R_4 = CH_2CH_3$
10 : $X = X' = Y' = O$; $Y = S$; $R_1 = R_2 = R_3 = R_4 = CH_2CH_3$
11 : $X = X' = Y = Y' = O$; $R_1 = R_2 = R_3 = CH_2CH_3$; $R_4 = H$
12 : $X = X' = Y' = O$; $Y = S$; $R_1 = R_2 = R_3 = CH_2CH_3$; $R_4 = H$
13 : $X = Y = S$; $X' = Y' = O$; $R_1 = R_2 = R_3 = CH_2CH_3$; $R_4 = H$
14 : $X = X' = Y = O$; $X' = Y' = S$; $R_1 = R_2 = R_3 = CH_2CH_3$; $R_4 = H$
15; 19: $X = Y = O$; $X' = Y' = S$; $R_1 = R_2 = R_3 = R_4 = CH_2CH_3$

a) X = Y = 0, $(C_2H_3)_2NCOCl$, KOH, Me₂SO, toluene, 8 °C, 15 min, 81%; X = 0, Y = S, $(C_2H_5)_2NCSCl$, t-BuOK, Me₂SO, toluene, 8 °C, 15 min, 85%; X = Y = S, $(C_2H_5)_2NCSSLi$, HMPA, toluene, 110 °C, 2 h, 87%; b) 0.6M HCl, water, dioxane, 30 °C, 20 - 30 min, 70 - 88%; c) X' = Y' = 0, $(C_2H_5)_2NCOCl$, KOH, toluene, 0 °C, 2.3 - 2.6 h, 52 - 69%; X' = Y' = 0, $R_1 = H$, R_2H_5NCO , pyridine, 40 °C, 48 - 72 h, 91 - 98%; $R_1 = H$, R_2H_5NCS , pyridine, 40 °C, 96 h, 98%; $R_1 = H$, R_2H_5NCS , pyridine, 40 °C, 96 h, 98%; $R_1 = H$, R_2H_5NCS , pyridine, 40 °C, 96 h, 98%; $R_2 = H$, R_3H_5NCS , acetone, 56 °C, 1 - 4.5 h, 72 - 89%; d) 0.6M HCl, water, dioxane, 60 °C, 2 h, 50 - 70%.

X = O; Y = X' = Y' = S; $R_1 = R_2 = R_3 = R_4 = CH_2CH_3$

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Scheme 2

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Table 1: ¹H NMR Chemical Shifts (8 in ppm, in CDCl₃) for Compounds 1' and 2 - 16.

	H4	H.S	1-5 H-6 H-6'	(7 11	110/0	Others
4.53 4.67 4.50 4.50 4.50 4.47 4.58 4.67 4.58	1	7.11	,	٠ <u>-</u> ـــ	C(CH3)2	
4.67 4.50 4.68 4.50 4.47 4.58 4.67 4.67	4.15	4.15	.15 4.15 3.68	3.68	1.48 1.30	7.75 (H-ortho), 7.30 (H-meta)
4.50 4.68 4.50 4.47 4.58 4.67 4.58	4.35	4.05	.05 3.86	3.86	1,46 1.22	3.91, 3.87 (4H, NCH ₂), 1.17, 1.15 (6H, CH ₃)
4.68 4.50 4.47 4.58 4.67 4.43 4.58	4.13	4.14	4.00	3.91	1.42 1.31 1.23 1.20	3.17 (4H, NCH ₂), 1.16, 1.10 (6H, CH ₃)
4.50 4.47 4.58 4.67 4.43 4.58	4.14	4.15	4.01	3.91	1.44 1.31 1.22 1.20	3.82, 3.64 (4H, NCH ₂), 1.16, 1.10 (6H, CH ₃)
4.47 4.58 4.67 4.43 4.58	4.34	4.16	4.00	3.91	1.47 1.33 1.24 1.24	3.93 (2H, NCH2), 3.65, 3.64 (2H, NCH2), 1.21, 1.19 (6H, CH3)
4.58 4.67 4.43 4.58	4.04	3.57	3.57	3.36	1.40 1.23	3.19 (4H, NCH ₂), 1.05 (6H, CH ₃)
4.67 4.43 4.58	4.15	3.56	3.85	3.66	1.48 1.28	3.78, 3.71, 3.42 (4H, NCH2), 1.21, 1.14 (6H, CH3)
4.43	4.30	3.67	3.79	3.57	1.44 1.23	3.92 (2H, NCH2), 3.78, 3.73 (2H, NCH2), 1.20, 1.17 (6H, CH3)
4.58	4.03	3.58	4.30	3.99	1.31 1.13	3.10 (8H, NCH ₂), 0.95, 0.91 (18H, CH ₃)
	4.15	3.71	4.38	4.08	1.39 1.20	3.76, 3.65 (4H, CSNCH ₂), 3.17 (4H, CONCH ₂), 1.13, 1.11 (6H,
						CSNCH ₂ CH ₃), 1.01 (6H, CONCH ₂ CH ₃)
11 5.82 4.51 5.13	4.05	3.66	4.39	4.05	1.43 1.24	3.17 (6H, NHCH2), 1.06, 1.04 (9H, CH3)
12 5.83 4.60 5.94	4.17	3.77	77 4.42	4.09	1.47 1.27	3.77, 3.42 (4H, CSNCH ₂), 3.16 (2H, CONHCH ₂), 1.20 (3H,
						CONHCH2CH3), 1.13, 1.07 (6H, CSNCH2CH3)
13 5.72 4.69 4.77	4.36	4.05	4.36	4.05	1.48 1.27	3.94, 3.69 (4H, CSNCH ₂), 3.12 (2H, COHNC <i>H</i> ₂), 1.22, 1.20 (6H,
						CSNCH ₂ CH ₃), 1.04 (3H, CONHCH ₂ CH ₃)
14 5.84 4.52 5.16	4.13	3.79	.79 4.83	4.35	1.45 1.26	3.50 (2H, CSNHCH2), 3.19 (4H, CONCH2), 1.17 (3H,
(Z)						CSNHCH ₂ CH ₃), 1.08 (6H, CONCH ₂ CH ₃)
14 5.84 4.52 5.16	4.13	3.79	.79 4.83	4.35	1.45 1.26	3.33 (2H, CSNHCH2), 3.19 (4H, CONCH2), 1.17 (3H,
(E)						CSNHCH ₂ CH ₃), 1.08 (6H, CONCH ₂ CH ₃)
15 5.71 4.40 4.99	3.97	3.75	.75 3.77	3.27	1.32 1.11	3.77, 3.61 (4H, CSNCH ₂), 3.07 (4H, CONCH ₂), 1.04 (6H,
						CONCH ₂ CH ₃), 0.93 (6H, C-6-SCSNCH ₂ CH ₃)
16 5.83 4.64 5.87	4.19	3.98	98 3.98 3.47	3.47	1.48 1.26	3.92, 3.72 (4H, C-6-SCSNCH2), 3.74, 3.40 (4H, C-3-OCSNCH2), 1.22,
						1.20 (6H, C-6-SCSNCH ₂ CH ₃), 1.11, 1.09 (6H, C-3-OCSNCH ₂ CH ₃)

Table 2: ¹H NMR Coupling Constants (in Hz) for compounds 1' and 2 - 16.

Compd	Coup	ling co	nstanc	es (Hz)			
	J _{1,2}	J _{2,3}	J _{3,4}	J _{4.5}	J _{5.6}	J _{5,6} ,	J _{6,6} ,	Others
1'	3.5	4.0				4.0	10.0	$J_{0,m} = 8.3$
2	3.9	0	3.6	8.8		7.1	14.4	
3	3.6	0	2.4	7.8	5.9	4.8	8.5	$J_{NCH_2-CH_3} = 7.1$
4	3.8	0	2.6		5.3	4.6	8.3	$J_{NCH_2-CH_3} = 7.0$
5	3.5	0	3.5	8.2	6.1	5.0	8.7	$J_{NCH_2-CH_3} = 6.9$
6	3.7	0	2.7	9.5	4.8	5.9	11.8	$J_{NCH_2-CH_3} = 7.0$
7	3.7	0	2.3	9.0	5.3	4.6	8.3	$J_{NCH_2} - C_{H_3} = 7.0$
8	3.6	0	3.5	8.9	3.5	5.9	11.4	$J_{NCH_2} - C_{H_3} = 7.1$
9	3.5	0	2.3	11.7	2.3	5.0	11.8	$J_{NCH_2} - C_{H_3} = 7.0$
10	3.7	0	2.6	9.3	2.7	4.9	11.9	$J_{NCH_2} - C_{H_3} = 7.1$
11	3.6	0	2.3					
12	3.5	0	3.7	9.0	2.2	5.0	12.0	
13	33	0	3.4					
14 (Z)	3.6	0	1.9	7.2		6.6	11.4	
14 (E)	3.6	0	1.9	7.2	1.7	6.6	11.4	$J_{NCH_2-CH_3} = 7.1$
15	3.6	0	2.2	8.9		7.5	13.9	
16	3.7	0	2.7	8.7		8.1	15.3	

Step b: Selective partial acid-catalyzed deprotection of derivatives 3-5 in 1,4-dioxane-H₂O-HCl at 30 °C afforded the intermediates 6-8 in yields ranging from 70 to 88%. Neither intramolecular transcarbamoylation nor decarbamoylation was observed in acidic (during the deprotection reaction), neutral and basic (during product extraction - see experimental section) medium. Such an observation is in contrast with that found for the analogous carboxylic acid esters. 16-18 or carbamoyl acid esters. 19-21 However, in the latter carbamoyl derivatives, product degradation was observed under strong basic 19-20 or strong acidic conditions 21 after 5 to 20 h.

The structures of the 3-carbamoyl ester derivatives of D-glucose **3-8** were supported by the ¹³C, ¹H NMR data and elemental analyses. In comparison with that of the parent sugar, the *O*-carbamoyl ester **6** showed a downfield shift for the resonance of H-3 (ca. +0.9 ppm), H-2 (ca. +0.1 ppm), H-4 (ca. +0.2 ppm) and C-3 (ca. +3.2 ppm) along with an upfield shift of the resonance of C-2 (ca. -2.0 ppm) and C-4 (ca. -1.8 ppm). These phenomena were amplified for the H-3 and C-3 signals of the *O*-thiocarbamoyl ester

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Table	3; ¹³ C	NMR	Chem	iical S	hifts (8 in pp	m, in CDCl ₃)	Table 3: ¹³ C NMR Chemical Shifts (δ in ppm, in CDCl ₃) for Compounds 1' and 2 - 13	nd 2 - 13.
Compd	Compd Chemical shifts (δ)	ical sh	ifts (δ)						
	C-1	C-1 C-2 C-3	C-3	C-4	C-5	C-5 C-6	C(CH3)2	C(CH3)2	Others
1,	103.0	103.0 82.6 18.1	18.1	81.5	69.4	69.4 70.7	111.9	26.6 26.5	144.9 (C-ipso), 132.6 (C-para), 129.9 (C-ortho), 127.9 (C-meta)
7	104.5	9.98	57.3	80.4		69.9 40.6	111.2	26.6 26.4	196.6 (C-3-SCS), 192.6 (C-6SCS), 50.0, 47.1 (2C, C-6-SCSN
									CH ₂), 49.7, 47.1 (2C, C-3-SCSNCH ₂), 12.5, 11.5 (4C, CH ₃)
3	104.8	83.5	77.0	80.0	72.6	72.6 67.2	112.0 109.1	26.6 26.6 26.1 25.1	154.3 (CO), 41.9, 41.2 (2C, NCH ₂), 13.9, 13.3 (2C, CH ₃)
4	104.7	83.5	79.9	82.2	72.4	67.4	112.1 109.3	27.7 26.6 26.1 25.1	185.5 (CS), 47.9, 43.3 (2C, NCH ₂), 13.2, 11.7 (2C, CH ₃)
ĸ	104.5	86.7	57.1	74.0	73.9	67.4	112.1 109.4	26.7 26.5 26.3 25.2	191.9 (CS), 49.3, 46.6 (2C, NCH ₂), 12.4, 11.4 (2C, CH ₃)
9	104.3	82.6	76.3	78.1	68.4	63.4	110.8	26.3 25.9	153.9 (CO), 41.3, 40.8 (2C, NCH2), 13.8, 13.2 (2C, CH3)
7	104.8	104.8, 83.2	82.0	80.3	67.5	64.2	112.4	26.6 26.2	185.8 (CS), 48.5, 43.9 (2C, NCH ₂), 13.1, 11.7 (2C, CH ₃)
œ	104.5	85.9	58.5		70.3	64.1	112.2	26.4 26.2	192.7 (CS), 50.2, 47.4 (2C, NCH ₂), 12.5, 11.4 (2C, CH ₃)
6	104.7	83.2	77.1	79.1	67.0	9.99	111.8	26.5 26.1	156.3 (C-6-0CO), 155.2 (C-3-0CO), 42.1, 41.5 (4C, NCH2),
									13.8, 12.1 (4C, CH ₃)
10	104.7	104.7 83.3	78.9	81.9	67.3	67.3 66.9	112.0	26.5 26.2	185.7 (CS), 156.6 (CO), 48.2, 43.6 (2C, CSNCH2), 41.8, 41.4
									(2C, CONCH2), 13.9, 13.1 (2C, CONCH2CH3), 13.1, 11.7 (2C,
									CSNCH ₂ CH ₃)
11	104.8	104.8 83.2	77.2	79.4		6.99 66.3	112.2	26.5 26.3	155.6 (C-6-0C0), 155.4 (C-3-0C0), 42.3, 41.6 (2C, C-3-
									OCONCH2), 35.8 (C-6-OCONHCH2), 15.0 (1C, C-6-OCO
									NHCH2CH3), 13.8, 13.2 (2C, C-3-0C0NCH2CH3)
12	104.7	83.4	82.0	79.2		67.0 66.5	112.3	26.6 26.2	185.8 (CS), 156.6 (CO), 48.3, 43.7 (2C, CSNCH2), 35.9 (1C,
									CONHCH ₂), 15.0 (1C, CONHCH ₂ CH ₃), 13.2, 11.7 (2C,
									CSNCH2CH3)
13	104.5	104.5 86.3	58.3	78.2		69.2 66.6	112.2	26.5 26.3	192.6 (CS), 156.7 (CO), 50.0, 47.2 (2C, CSNCH2), 35.9, 35.1
									(2C, CONHCH ₂), 15.7, 15.0 (2C, CONHCH ₂ CH ₃), 12.5, 11.3
									(2C, CSNCH2CH3)

Table 4: ^{13}C NMR Chemical Shifts (δ in ppm, in CDCl₃) for Compounds 14 - 19.

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Compd	Comnd Chemical shifts (8)	ical sh	iAs (8)							
	C-I	C-2	C-3	C.4	C-5	C-6	C-5 C-6 C(CH ₃) ₂ C(CH ₃) ₂ Others	C.E.	13)2	Others
14	104.8	83.2	77.1	9.62	71.3	1.99	71.3 66.7 112.2	26.6	26.2	26.6 26.2 190.0 (CS), 155.5 (CO), 42.4, 41.7 (2C, CONCH2), 40.1 (1C, CSNHCH2),
(Z)										13.8, 13.2 (2C, CONCH2CH3), 13.6 (1C, CSNHCH2CH3)
14	104.8	83.2	77.1	9.62	71.6 66.5	66.5	112.2	26.6 26.2		190.0 (CS), 155.5 (CO), 42.4, 41.7 (2C, CONCH2), 38.1 (1C, CSNCH2),
(E)										14.2 (IC, CSNCH2CH3), 13.8, 13.2 (2C, CONCH2CH3),
15	104.7	83.2	8.9/	81.7	67.3 40.9	40.9	111.9	26.6 26.1		195.9 (CS), 155.0 (CO), 49.6, 46.8 (2C, CSNCH ₂), 42.2, 41.5 (2C,
									_	CONCH ₂), 13.8, 13.1 (2C, CONCH ₂ CH ₃), 12.3, 11.3 (2C, CSNCH ₂ CH ₃)
16	104.7	83.5	81.5	81.9	67.7 40.6	40.6	112.2	26.7 26.3	26.3	196.4 (SCS), 185.7 (OCS), 49.9, 47.0 (2C, SCSNCH2), 48.1, 43.7 (2C,
									-	OCSNCH ₂), 13.3, 11.7 (2C, OCSNCH ₂ CH ₃), 12.4, 11.5 (2C,
										SCSNCH ₂ (H ₃)
17^{a}	7.16	71.2	57.3	70.7	70.9 40.5	40.5				195.9 (C-6-SCS), 194.3 (C-3-SCS), 49.3, 46.9 (2C, C-6-SCSNCH ₂),
8									-	48.9, 46.4 (2C, C-3-SCSNCH ₂), 12.3, 11.4 (4C, CH ₃)
17	98.4	72.3	8.09	71.8	76.4 40.5	40.5				195.2 (C-6-SCS), 194.2 (C-3-SCS), 49.3, 46.9 (2C, C-6-SCSNCH ₂),
(b)									•	48.9, 46.4 (2C, C-3-SCSNCH ₂), 12.3, 11.4 (4C, CH ₃)
18	92.1	70.6	76.1	68.7	69.5 63.9	63.9				155.6 (C-6-0CO), 154.9 (C-3-0CO), 40.9 (4C, NCH ₂), 13.7 (4C, CH ₃)
(8)										
183	2.96	73.0	78.4	68.7	73.6 64.0	64.0				155.3 (2C, CO), 40.9 (4C, NCH ₂), 13.7 (4C, CH ₃)
<u>@</u>										
19 4	92.2	71.9	76.0	69.4	71.7 40.3	40.3				194.3 (CS), 155.3 (CO), 48.8, 46.4 (2C, CSNCH ₂), 41.0 (2C,
(δ)									-	CONCH ₂), 13.6 (2C, CONCH ₂ CH ₃), 12.3, 11.3 (2C, CSNCH ₂ CH ₃)
19ª	8.96	73.1 78.3	78.3	9.07	73.6 41.3	41.3				194.3 (CS), 155.6 (CO), 48.8, 46.4 (2C, CSNCH2), 41.0 (2C,
(θ)									-	CONCH ₂), 13.6 (2C, CONCH ₂ CH ₃), 12.3, 11.3 (2C, CSNCH ₂ CH ₃)
S. C.	Spectra were recorded in (CD)	Pro rec	babao) ui	057.0					

a. Spectra were recorded in (CD₃)₂SO.

7 having a thiocarbonyl group such as H-3 (ca. +2.0 ppm), H-2 (ca. +0.2 ppm), H-4 (ca. +0.3 ppm), C-3 (ca. +8.9 ppm), C-2 (ca. -1.4 ppm) and C-4 (ca. +0.4 ppm)). The S-thiocarbamoyl ester 8 showed a downfield shift for the resonance of H-3 (ca. +0.7 ppm), H-2 (ca. +0.3 ppm), H-4 (ca. +0.5 ppm) and C-2 (ca. +1.3 ppm) along with an upfield shift of the resonance of C-3 (ca. -14.6 ppm) and C-4 (ca. -1.1 ppm). The proton and carbon signals of 3-carbamoyl ester derivatives of D-glucose 3-5 were similar to those described above.

Step c: Regioselective N,N-diethylcarbamoylation at C-6 of compounds 6 and 7 with N,N-diethylcarbamoyl chloride (1.0 equiv) in the presence of finely powdered KOH (2.0 equiv) in toluene at 0 °C gave the bis-3,6-O-(N,N-diethylcarbamoyl) esters 9 (2.6 h in 69% yield) and the 6-O-(N,N-diethylcarbamoyl)-3-O-(N,N-diethylthiocarbamoyl) esters 10 (2.3 h in 52% yield) respectively. Each of these reactions gave two by-products in small amounts due to competitive reactions during carbamoylation of compounds 6 and 7 (Scheme 3) respectively. Thus the carbamoylation of 7 gave:

- (i) bis-5,6-O-(N,N-diethylcarbamoyl)-3-O-(N,N-diethylthiocarbamoyl)-1,2-O-iso-propylidene α-D-glucofuranose (10°) by carbamoylation at C-5 and C-6 sites;
- (ii) bis-3,6-O-(N,N-diethylcarbamoyl)-5-O-(N,N-diethylthiocarbamoyl)-1,2-O-iso-propylidene-α-D-glucofuranose (10'') by internal transthiocarbamoylation from C-3 to C-5 position followed by reaction of the liberated C-3 alcoholate with N,N-diethylcarbamoyl chloride. N,N-diethylcarbamoylation of thiocarbamate derivative 7 allowed us to verify the transesterification phenomenon since the diester 10 gave the triester 10'', in basic media, by migration of the thioamide group from C-3 to C-5 position through a cyclic intermediate.

Similary, the by-products obtained from the carbamoylation of 6 were both tris-3,5,6-O-(N,N-diethylcarbamoyl)-1,2-O-isopropylidene- α -D-glucofuranose (9°) and bis-5,6-O-(N,N-diethylcarbamoyl)-1,2-O-isopropylidene- α -D-glucofuranose (9°°).

Regiospecific N-ethylcarbamoylation at C-6 of compounds 6-8 with ethyl isocyanate and ethyl isothiocyanate (2.0 equiv) in pyridine at 40 °C gave the 6-O-(N-ethyl carbamate) derivatives 11-13 (48-72 h in 91-98% yield) and the 6-O-(N-ethyl thiocarbamate) derivatives 14 (96 h in 98% yield) respectively. We did not detect any other product in the reaction mixture indicating that neither C-5 carbamoylation, nor transcarbamoylation from C-3 to C-5 took place. It is important to note the feature

(C₂H₅) ₂NCOCl, KOH, toluene, 0 °C, 2.3 h.

Scheme 3

distinguishing the N-alkyl carbamate and the N-alkyl thiocarbamate. It is known that there is a high barrier to rotation around the N-CS bond. We also observed this phenomenon with the N-ethylthiocarbamoyl ester 14 using NMR spectroscopy but not with the N-ethylcarbamoyl ester 11 due to the free rotation of the carbamoyl bond. The methylene proton signals assigned to NCH₂ of the thiocarbamoyl group are indicative of the Z and E conformations to compound 14 because the thiocarbonyl group anisotropy caused deshielding of CH₂ in the Z conformer (ca. +0.17 ppm).

Regiospecific N,N-diethyldithiocarbamoylation at C-6 of compounds 6-8 was performed by C-6 tosylation²⁵ followed by substitution with lithium N,N-diethyl dithiocarbamate in acetone at 56 °C, to give the 6-S-(N,N-diethyl dithiocarbamate) derivatives 15, 16 and 2 (1-4.5 h in 72-89% yield starting from the tosylate precursor). It is interesting to observe that compound 2 is obtained in 28% overall yield from compound 1 using the pathway shown in Scheme 2 instead of 22% as outlined in Scheme 1 which involves the intermediate. The route shown in Scheme 2 has the advantage that it permits convenient access to a large range of products having identical or different carbamoyl

groups at the C-3 and C-6 positions whereas the route outlined in Scheme 1 led only to compounds with the same carbamoyl group in each position.

Step d: The biscarbamate derivatives 2, 9 and 15 were each deprotected to complete the product range and to extend the hydrophilic-lipophilic balance. This step was performed using 1,4-dioxane-H₂O-HCl at 60 °C (instead of 30 °C which was used to prepare partially deprotected compounds 3-5) to give 17-19 respectively (2 h in 50-70% yield).

Antifungal properties. Of the bis(carbamoyl esters) tested using a solid medium (Czapek Yeast Agar, glycine, agar²⁶) and a Fayret liquid medium (glucose, ammonium nitrate²⁷), antifungal activities were observed for compounds 2 and 17; each had the same bis(N,N-diethyldithiocarbamoyl group) at C-3 and C-6 but differed by having: (i) one or three hydroxyl groups free (corresponding to the presence or absence of an isopropylidene group respectively); (ii) furanosyl or pyranosyl forms respectively. Compound 2 inhibited sporulation of Botrytis cinerea (no-operating conidiophore formation after 4 days at 100 ppm) and mycellium growth of Fusarium oxysporum f.sp. lini (mycellium growth inhibition: 17% after 4 days at 200 ppm). 17 inhibited mycellium growth of Botrytis cinerea (mycellium growth inhibition: 50% after 4 days at 200 ppm) and Fusarium oxysporum f.sp. lini (mycellium growth inhibition: 18% after 4 days at 200 ppm). These results showed that the type of saccharide moiety was important for biological activity since the bis(dithiocarbamoyl esters) 2 and 17 had selective antifungal activities at different stages of the fungal growth.

EXPERIMENTAL

General Procedure. Melting points were determined on a digital melting-point apparatus (Electrothermal) and are uncorrected. Optical rotations were recorded at 22 °C in CHCl₃ or MeOH solutions with a digital polarimeter DIP-370 (JASCO) using a 1dm cell. ¹H NMR and ¹³C NMR spectra were recorded in CDCl₃ or Me₂SO-d₆ (internal Me₄Si) respectively at 300.13 MHz and at 75.47 MHz (Brüker AM WB-300). TLC was performed on Silica F254 (Merck) and detection by UV light at 254 nm or by charring with phosphomolybdic-H₂SO₄ reagent. Column chromatography was effected on Silica Gel 60 (Merck, 230 mesh). Me₂CO, hexane, ether and each industrial grade were supplied

by CINAS. Elemental analyses were performed by the Service Central de Micro-Analyse du Centre National de Recherche Scientifique (Vernaison, France). 1,2:5,6-Di-O-isopropylidene-α-D-glucofuranose (1), N,N-diethylcarbamoyl chloride, ethyl isocyanate, ethyl isothiocyanate, bases and solvents were supplied by JANSSEN or ALDRICH.

N,N-Diethyldithiocarbamic acid lithium salt was synthesized in accordance with the method described by Postel. 15

3-Deoxy-3-iodo-1,2:5,6-di-*O*-isopropylidene-α-D-allofuranose was synthesized in accordance with the method described by Garegg. ¹⁴

3-Deoxy-3-iodo-1,2-*O*-isopropylidene-6-*O*-tosyl-α-D-allofuranose (1'). 3-Deoxy-3-iodo-1,2:5,6-di-*O*-isopropylidene-α-D-allofuranose (7.0 g, 18.9 mmol) was reacted with a 0.6M HCl in water-dioxane (7:93, 65 mL) at 30 °C during 30 min. After addition of sodium hydrogen carbonate (pH 7), the mixture was stirred for 10 min, filtered and the solvent was removed under reduced pressure. The crude product was purified on a silica gel column eluted with hexane-acetone (1:1) to give 4.0 g (64%) of 3-deoxy-3-iodo-1,2-*O*-isopropylidene-α-D-allofuranose, mp 110-111 °C; $[\alpha]_D^{22}$ +113° (*c* 1.2, CHCl₃); ¹H NMR (CDCl₃) δ 5.76 (d, 1H, J_{1,2} = 3.6 Hz, H-1), 4.57 (t, 1H, J_{2,3} = 4.2, H-2), 4.25 (dd, 1H, J_{4,5} = 2.9 Hz, H-4), 4.00 (m, 1H, J_{5,6} = 7.2 Hz, H-5), 3.88 (dd, 1H, J_{3,4} = 10.2 Hz, H-3), 3.83 (dd, 1H, J_{6,6} = 11.5 Hz, H-6), 3.68 (dd, 1H, J_{5,6} = 3.9 Hz, H-6'), 1.45, 1.27 (s, 6H, C(CH₃)₂); ¹³C NMR (CDCl₃) δ 111.8 (*C*(CH₃)₂), 103.0 (C-1), 83.5 (C-2), 81.5 (C-4), 71.5 (C-5), 63.4 (C-6), 26.5 (2C, C(*C*(H₃)₂), 18.7 (C-3).

p-Toluenesulphonyl chloride (4.0 g, 21.2 mmol) dissolved in toluene (27 mL) was slowly added to a stirred pyridine (27 mL) solution of 3-deoxy-3-iodo-1,2-O-isopropylidene- α -D-allofuranose (7.0 g, 21.2 mmol) at 0 °C. After 48 h at 4 °C, crushed ice and aqueous-HCl (9.1) (50 mL) were added to the mixture and two phases separated. The aqueous phase was extracted with toluene (2x25 mL); the organic phases were pooled, dried (Na₂SO₄) and concentrated. The crude product was purified on a silica gel column eluted with hexane-acetone (4:1) to give 8.6 g (84%) of 1°. [α]_D²² +233° (c 1.2, CHCl₃).

Anal. Calcd for $C_{16}H_{21}O_7SI$ (484.31): C, 39.68; H, 4.37, S, 6.62. Found: C, 39.42; H, 4.54, S, 7.01.

3,6-Dideoxy-bis-3,6-S-(N,N-diethyldithiocarbamoyl)-1,2-O-isopropylidene- α -D-glucofuranose (2).

Method α : N,N-Diethyldithiocarbamic acid lithium salt (1.9 g, 12.4 mmol) was added dropwise to a stirred solution of 1° (1.5 g, 3.1 mmol) in HMPA-toluene (1:1) (5 mL). Upon completion of the addition, the reaction mixture was stirred at 110 °C for 24 h. The mixture was filtered and the solvent removed by distillation under reduced pressure to yield a viscous oil. This oily residue was treated with hexane-diethyl ether (1:1) and washed twice with water and dried (Na₂SO₄). The crude product was purified on a silica gel column eluted with hexane-acetone (9:1) to give 0.9 g (60%) of 2. $[\alpha]_D^{22}$ +48° (c 1.1, CHCl₃).

Anal. Calcd for $C_{19}H_{34}O_4S_4N_2$ (482.75): C, 47.27; H, 7.10; S, 26.57; N, 5.80. Found: C, 47.21; H, 7.02; S, 27.02; N, 5.82.

Method b: p-Toluenesulphonyl chloride (3.8 g, 19.9 mmol) dissolved in toluene (25 mL) was slowly added to a stirred pyridine (25 mL) solution of 8 (7.0 g, 19.9 mmol) at 0 °C. After 48 h at 4 °C, crushed ice and aqueous-HCl (9:1) (50 mL) were added to the mixture and two phases separated. The aqueous phase was extracted with toluene (2x25 mL); the organic phases were pooled, dried (Na₂SO₄) and concentrated. The crude product was purified on a silica gel column eluted with hexane-diethyl ether (4:1) to give 8.8 g (87% yield) of 3-deoxy-3-S-(N,N-diethyldithiocarbamoyl)-1,2-O-isopropylidene-6-O-tosyl-α-D-glucofuranose (8'). $[\alpha]_D^{22}$ +76° (c 2.0, CHCl₃); ¹H NMR (CDCl₃) δ 7.75 (d, 2H, $J_{0,m} = 8.3$ Hz, H-ortho), 7.27 (d, 2H, H-meta), 5.69 (d, 1H, $J_{1,2} = 3.5$ Hz, H-1), 4.84 (d, 1H, $J_{3,4} = 3.4$ Hz, H-3), 4.71 (d, 1H, $J_{2,3} = 0$, H-2), 4.27 (dd, 2H, H-6, H-6'), 4.00 (dd, 1H, $J_{4,5} = 11.7$ Hz, H-4), 3.92 (m, 1H, H-5), 4.00, 3.70 (q, 4H, NCH₂), 2.38 (s, 3H, CH₃), 1.46, 1.25 (s, 6H, C(CH₃)₂), 1.25, 1.22 (t, 6H, J = 7.0 Hz, CH₃); 13 C NMR (CDCl₃) δ 192.4 (CS), 144.6 (C-ipso), 132.9 (C-para), 129.7 (C-ortho), 128.0 (C-meta), 112.2 $(C(CH_3)_2)$, 104.5 (C-1), 85.9 (C-2), 78.1 (C-4), 71.8 (C-5), 68.7 (C-6), 58.5 (C-3), 50.4, 47.4 (2C, NCH₂), 26.4, 26.2 (2C, C(CH₃)₂), 21.5 (CH₃), 12.5, 11.4 (2C, NCH₂CH₃). N,N-Diethyldithiocarbamic acid lithium salt (2.3 g, 14.8 mmol) and 8' (5.0 g, 9.9 mmol) in acetone (40 mL) was stirred at 56 °C for 4.5 h. The mixture was filtered and acetone was removed by distillation under reduced pressure to yield a viscous oil. This oily residue was treated with hexane-diethyl ether (1:1) and washed twice with water and dried (Na₂SO₄).

The crude product was purified on a silica gel column eluted with hexane-acetone (9:1) to give 3.4 g (72%) of 2.

3-O-(N,N-Diethylcarbamoyl)-1,2:5,6-di-O-isopropylidene- α -D-glucofuranose (3). N,N-Diethylcarbamoyl chloride (3.1 g, 22.9 mmol) was added dropwise to a stirred solution of 1 (5.0 g, 19.2 mmol) and KOH (2.2 g, 38.5 mmol) in Me₂SO-toluene (5:95, 50 mL) at 8 °C for 15 min. Ammonium chloride in aqueous solution was added and the mixture stirred for a further 10 min. The organic extract was dried (Na₂SO₄) and the solvent was evaporated under reduced pressure. The crude product was purified on a silica gel column eluted with hexane-acetone (9:1) to give 5.6 g (81%) of 3. $[\alpha]_D^{22}$ -38° (c 1.5, CHCl₃).

Anal. Calcd for $C_{17}H_{29}O_7N$ (359.42): C, 56.81; H, 8.13; N, 3.90. Found: C, 56.86; H, 8.15; N, 4.11.

3-O-(N,N-Diethylthiocarbamoyl)-1,2:5,6-di-O-isopropylidene- α -D-glucofuranose (4). To a stirred solution of thiophosgene (2.9 g, 25.2 mmol) in toluene (27 mL) at 8 °C was added diethylamine (3.7 g, 50.4 mmol) dropwise in toluene (8 mL). At the end of the addition, the diethylammonium chloride was removed by filtration. The filtrate (26 mL) was added dropwise to a stirred mixture of 1 (5.0 g, 19.2 mmol) and potassium t-butoxide (2.6 g, 23.0 mmol) in Me₂SO-toluene (13:87, 15 mL) below 8 °C for 15 min. Ammonium chloride in aqueous solution was added and the mixture stirred for a further 10 min. The organic extract was dried (Na₂SO₄) and the solvent was evaporated under reduced pressure. The crude product was purified on a silica gel column eluted with hexane-acetone (9:1) to give 6.2 (85%) of 4, mp 53-54 °C; $[\alpha]_D^{22}$ -69° (c 1.6, CHCl₃).

Anal. Calcd for $C_{17}H_{29}O_6SN$ (375.49): C, 54.38; H, 7.78; S, 8.54; N, 3.73. Found: C, 54.10; H, 7.71; S, 8.35; N, 3.46.

3-Deoxy-3-S-(N,N-diethyldithiocarbamoyl)-1,2:5,6-di-O-isopropylidene-α-D-glucofuranose (5). A solution of 3-deoxy-3-iodo-1,2:5,6-di-O-isopropylidene-α-D-allofuranose (10.0 g, 27.0 mmol) and N,N-diethyldithiocarbamic acid lithium salt (6.3 g, 40.5 mmol) in HMPA-toluene (1:1) (5 mL) was stirred at 110 °C for 2 h. The mixture was cooled at room temperature and a solution of hexane-water (1:1) (200 mL) was added. The aqueous phase was extracted with hexane-diethyl ether (1:1) (100 mL), the organic phases were washed twice with water and dried (Na₂SO₄). The crude product was purified

on a silica gel column eluted with hexane-acetone (9:1) to give 9.2 g (87%) of 5, mp 83-84 °C; $[\alpha]_D^{22}$ -44° (c 1.5, CHCl₃).

Anal. Calcd for $C_{17}H_{29}O_5S_2N$ (391.55): C, 52.15; H, 7.46; S, 16.38; N, 3.58. Found: C, 52.21; H, 7.55; S, 16.60; N, 3.62.

3-O-(N,N-Diethylcarbamoyl)-1,2-O-isopropylidene- α -D-glucofuranose (6). 3 (7.0 g, 19.5 mmol) was reacted with a 0.6M HCl in water-dioxane (1:9) (200 mL) at 30 ° C for 20 min. After addition of sodium hydrogen carbonate (pH 6), the mixture was stirred 10 min, filtered and the solvent was removed under reduced pressure. The crude product was purified on a silica gel column eluted with hexane-acetone (1:1) to give 4.3 g (70%) of 6. $[\alpha]_D^{22}$ +64° (c 1.4, CHCl₃).

Anal. Calcd for $C_{14}H_{25}O_7N$ (319.35): C, 52.65; H, 7.89; N, 4.39. Found: C, 52.52; H, 8.03; N, 4.17.

3-O-(N,N-Diethylthiocarbamoyl)-1,2-O-isopropylidene- α -D-glucofuranose (7). Likewise, 4 (7.0 g, 18.7 mmol) gave, after 30 min, 5.5 g (88%) of 7, mp 84-85 °C; $[\alpha]_D^{22}$ +64° (c 1.4, CHCl₃).

Anal. Calcd for $C_{14}H_{25}O_6SN$ (335.42): C, 50.13; H, 7.51; S, 9.56; N, 4.18. Found: C, 50.20; H, 7.63; S, 10.05; N, 4.31.

3-Deoxy-3-S-(N,N-diethyldithiocarbamoyl)-1,2-O-isopropylidene- α -D-gluco-furanose (8). Likewise, 5 (10.0 g, 25.6 mmol) gave, after 30 min, 6.8 g (75%) of 8, mp 78-80 °C; $[\alpha]_D^{22}$ +91° (c 1.2, CHCl₃).

Anal. Calcd for $C_{14}H_{25}O_5S_2N$ (351.49): C, 47.84; H, 7.17; S, 18.25; N, 3.98. Found: C, 47.76; H, 7.03; S, 18.95; N, 4.02.

Bis-3,6-O-(N,N-diethylcarbamoyl)-1,2-O-isopropylidene- α -D-glucofuranose

(9). A solution of N,N-diethylcarbamoyl chloride (0.8 g, 6.3 mmol) in toluene (4 mL) was added dropwise to a stirred solution of 6 (2.0 g, 6.3 mmol) and KOH (0.7 g, 12.5 mmol) in toluene (20 mL) at 0 °C. Stirring was continued at 0 °C until HPLC monitoring indicated that maximum conversion was achieved (2.6 h), whereupon a saturated aqueous solution of ammonium chloride was added and the mixture stirred for a further 10 min. The organic extract was dried (Na₂SO₄) and the solvent was evaporated under reduced pressure. The crude product was purified on a silica gel column eluted with hexane-diethyl ether (4:1) to give 0.2 g of a mixture containing tris-3,5,6-O-(N,N-diethylcarbamoyl)-1,2-

O-isopropylidene-α-D-glucofuranose (9') (95% estimated by 1 H and 13 C NMR integration). 1 H NMR (CDCl₃) δ 5.69 (d, 1H, J_{1,2} = 3.6 Hz, H-1), 5.00 (d, 1H, J_{3,4} = 3.0 Hz, H-3), 4.91 (dq, 1H, J_{5.6} = 2.2 Hz, H-5), 4.37 (dd, 1H, J_{5.6} = 4.1 Hz, H-6), 4.34 (d, 1H, J_{2,3} = 0, H-2), 4.30 (dd, 1H, J_{4,5} = 9.4 Hz, H-4), 3.97 (dd, 1H, J_{6.6} = 12.1 Hz, H-6'), 3.05 (m, 12H, NCH₂), 1.30, 1.10 (s, 6H, C(CH₃)₂), 0.89, 0.86 (t, 18H, J = 7.2 Hz, CH₃); 13 C NMR (CDCl₃) δ 155.2 (C-6-OCO), 153.8 (2C, C-3-OCO, C-5-OCO), 111.9 (C(CH₃)₂), 104.8 (C-1), 83.3 (C-2), 76.8 (C-4), 76.0 (C-3), 68.9 (C-5), 64.0 (C-6), 41.8, 41.1 (6C, NCH₂), 26.6, 26.0 (2C, C(CH₃)₂), 13.8, 13.2 (6C, CH₃).

Further elution provided 0.1 g of a mixture containing bis-5,6-O-(N,N-diethylcarbamoyl)-1,2-O-isopropylidene- α -D-glucofuranose (9°) (94% estimated by 1 H and 13 C NMR integration). 1 H NMR (CDCl₃) δ 5.81 (d, 1H, J_{1,2} = 3.5 Hz, H-1), 4.92 (m, 1H, H-5), 4.45 (dd, 1H, J_{5,6} = 2.3 Hz, H-6), 4.44 (d, 1H, J_{2,3} = 0, H-2), 4.26 (dd, 1H, J_{5,6} = 5.0 Hz, J_{6,6} = 11.8 Hz, H-6'), 4.01 (dd, 1H, J_{4,5} = 11.7 Hz, H-4), 3.96 (d, 1H, J_{3,4} = 2.3 Hz, H-3), 3.16 (m, 8H, NCH₂), 1.38, 1.21 (s, 6H, C(CH₃)₂), 1.01, 1.00 (t, 18H, J = 7.0 Hz, CH₃); 13 C NMR (CDCl₃) δ 156.1 (C-6-OCO), 155.3 (C-5-OCO), 111.5 (C(CH₃)₂), 105.1 (C-1), 84.4 (C-2), 79.3 (C-4), 73.6 (C-3), 70.2 (C-5), 64.4 (C-6), 42.2, 41.5 (4C, NCH₂), 26.7, 26.0 (2C, C(CH₃)₂), 13.7, 13.1 (4C, CH₃). Further elution provided 1.8 g (69%) of 9, mp 50-52 °C, [α]_D²² +15° (C 1.1, CHCl₃).

Anal. Calcd for $C_{19}H_{34}O_8N_2$ (418.49): C, 54.53; H, 8.19; N, 6.69. Found: C, 54.71; H, 8.32; N, 6.57.

Further elution provided 0.5 g of the initial product 6.

6-*O*-(*N*,*N*-Diethylcarbamoyl)-3-*O*-(*N*,*N*-diethylthiocarbamoyl)-1,2-*O*-isopropylidene-α-D-glucofuranose (10). Likewise, 7 (5.0 g, 14.9 mmol) and *N*,*N*-diethylcarbamoyl chloride (2.0 g, 14.9 mmol) gave, after 2.3 h, 0.4 g of a mixture containing bis-5,6-*O*-(*N*,*N*-diethylcarbamoyl)-3-*O*-(*N*,*N*-diethylthiocarbamoyl)-1,2-*O*-isopropylidene-α-D-glucofuranose (10') (96% estimated by ¹H and ¹³C NMR integration). ¹H NMR (CDCl₃) δ 5.76 (d, 1H, $J_{3,4} = 2.8$ Hz, H-3), 5.73 (d, 1H, $J_{1,2} = 3.5$ Hz, H-1), 4.97 (m, 1H, $J_{5.6} = 2.2$ Hz, H-5), 4.48 (d, 1H, $J_{2,3} = 0$, H-2), 4.42 (dd, 1H, $J_{5.6} = 4.2$ Hz, H-6), 4.41 (dd, 1H, $J_{4.5} = 9.3$ Hz, H-4), 4.03 (dd, 1H, $J_{6,6} = 12.1$ Hz, H-6'), 3.75, 3.65, 3.46, 3.00 (m, 4H, C-3-OCSNCH₂), 3.10 (m, 8H, C-5-OCONCH₂), C-6-OCONCH₂), 1.37, 1.16

(s, 6H, C(CH₃)₂), 1.05, 1.00 (t, 6H, J = 7.2 Hz, C-3-OCSNCH₂CH₃), 0.95 (t, 12H, J = 7.2 Hz, C-5-OCONCH₂CH₃, C-6-OCONCH₂CH₃); ¹³C NMR (CDCl₃) δ 185.0 (CS), 155.2 (C-6-OCO), 153.7 (C-5-OCO), 112.1 (C(CH₃)₂), 104.9 (C-1), 83.3 (C-2), 81.2 (C-4), 76.6 (C-3), 68.8 (C-5), 64.0 (C-6), 47.9, 43.8 (2C, C-3-OCSNCH₂), 43.8, 41.3 (4C, C-5-OCONCH₂, C-6-OCONCH₂), 26.6, 26.1 (2C, (C(CH₃)₂), 13.9, 13.1 (4C, C-5-OCONCH₂CH₃), C-6-OCONCH₂CH₃), 13.1, 11.6 (2C, C-3-OCSNCH₂CH₃).

Further elution provided 0.2 g of a mixture containing bis-3,6-O-(N,N-diethylcarbamoyl)-5-O-(N,N-diethylthiocarbamoyl)-1,2-O-isopropylidene- α -D-glucofuranose (10°) (95% estimated by 1 H and 13 C NMR integration). 1 H NMR (CDCl₃) δ 5.84 (d, 1H, J_{1,2} = 3.6 Hz, H-1), 5.20 (m, 1H, H-5), 5.16 (d, 1H, J_{3,4} = 2.9 Hz, H-3), 4.87 (dd, 1H, J_{4,5} = 11.9 Hz, H-4), 4.49 (d, 1H, J_{2,3} = 0, H-2), 4.46 (dd, 1H, J_{5,6} = 2.2 Hz, J_{6,6}· = 12.1 Hz, H-6), 4.42 (dd, 1H, J_{5,6}· = 4.2 Hz, H-6'), 3.74, 3.42, 3.10 (m, 4H, C-5-OCSNCH₂), 3.18 (m, 8H, C-3-OCONCH₂, C-6-OCONCH₂), 1.45, 1.25 (s, 6H, C(CH₃)₂), 1.03 (m, 18H, CH₃); 13 C NMR (CDCl₃) δ 185.0 (CS), 156.0 (2C, CO), 112.0 (C(CH₃)₂), 107.7 (C-1), 83.5 (C-2), 77.3 (C-4), 76.1 (C-3), 70.4 (C-5), 68.8 (C-6), 47.8, 43.7 (2C, C-5-OCSNCH₂), 2x42.2, 42.0, 41.3 (4C, C-3-OCONCH₂, C-6-OCONCH₂), 26.7, 26.1 (2C, C(CH₃)₂), 13.9, 13.2 (4C, C-3-OCONCH₂CH₃, C-6-OCONCH₂CH₃), 13.2, 11.7 (2C, C-5-OCSNCH₂CH₃). Further elution provided 3.4 g (52%) of 10. [α]_D²² +53° (c 1.2, CHCl₃).

Anal. Calcd for $C_{19}H_{34}O_7SN_2$ (434.55): C, 52.52; H, 7.89; S, 7.38; N, 6.45. Found: C, 52.75; H, 7.81; S, 7.59; N, 6.54.

Further elution provided 1.5 g of the initial product 7.

3-O-(N,N-Diethylcarbamoyl)-6-O-(N-ethylcarbamoyl)-1,2-O-isopropylidene- α -D-glucofuranose (11). A solution of 6 (2.0 g, 6.3 mmol) and ethyl isocyanate (0.9 g, 12.5 mmol) in pyridine (20 mL) was stirred at 40 °C for 48 h. The organic mixture was concentrated under reduced pressure. The crude product was purified on a silica gel column eluted with hexane-acetone (9:1) to give 2.4 g (98%) of 11. $[\alpha]_D^{22}$ +36° (c 1.1, CHCl₃).

Anal. Calcd for $C_{17}H_{30}O_8N_2$ (390.43): C, 52.30; H, 7.74; N, 7.17. Found: C, 52.00; H, 7.52; N, 7.13.

3-O-(N,N-Diethylthiocarbamoyl)-6-O-(N-ethylcarbamoyl)-1,2-O-isopropylidene- α -D-glucofuranose (12). Likewise, 7 (1.0 g, 3.0 mmol) and ethyl isocyanate (0.4 g, 6.0 mmol) gave, after 72 h, 1.1 g (91%) of 12. $[\alpha]_D^{22}$ +64° (c 1.5, CHCl₃).

Anal. Calcd for $C_{17}H_{30}O_7SN_2$ (406.50): C, 50.23; H, 7.44; S, 7.89; N, 6.89. Found: C, 50.35; H, 7.51; S, 8.11; N, 6.81.

3-Deoxy-3-S-(N,N-diethyldithiocarbamoyl)-6-O-(N-ethylcarbamoyl)-1,2-O-isopropylidene- α -D-glucofuranose (13). Likewise, 8 (2.0 g, 5.7 mmol) and ethyl isocyanate (0.4 g, 6.0 mmol) gave, after 72 h, 2.3 g (94%) of 13. $[\alpha]_D^{22}$ +64° (c 1.1, CHCl₃).

Anal. Calcd for $C_{17}H_{30}O_6S_2N_2$ (422.57): C, 48.32; H, 7.16; S, 15.18; N, 6.63. Found: C, 48.51; H, 7.25; S, 15.35; N, 6.58.

3-O-(N,N-Diethylcarbamoyl)-6-O-(N-ethylthiocarbamoyl)-1,2-O-isopropylidene- α -D-glucofuranose (14). Likewise, 6 (2.0 g, 6.3 mmol) and ethyl isothiocyanate (1.1 g, 12.5 mmol) in pyridine (5 mL) gave, after 96 h, 2.5 g (98%) of 14. $[\alpha]_D^{22}$ +45° (c 1.2, CHCl₃).

Anal. Calcd for $C_{17}H_{30}O_7SN_2$ (406.50): C, 50.23; H, 7.44; S, 7.89; N, 6.89. Found: C, 50.07; H, 7.53; S, 7.91; N, 6.73.

6-Deoxy-3-*O*-(*N*,*N*-diethylcarbamoyl)-6-*S*-(*N*,*N*-diethyldithiocarbamoyl)-1,2-*O*-isopropylidene-α-D-glucofuranose (15). The two-step procedure for the preparation of **2** (*method b*) was applied. In the first step (*method b*), **6** (5.0 g, 15.7 mmol) gave 5.9 g (80%) of 3-*O*-(*N*,*N*-diethylcarbamoyl)-1,2-*O*-isopropylidene-6-*O*-tosyl-α-D-glucofuranose (**6**'). [α]_D²² +41° (*c* 1.0, CHCl₃); ¹H NMR (CDCl₃) δ 7.76 (d, 2H, J_{o,m} = 8.2 Hz, H-ortho), 7.27 (d, 2H, H-meta), 5.80 (d, 1H, J_{1,2} = 3.6 Hz, H-1), 5.10 (d, 1H, J_{3,4} = 2.1 Hz, H-3), 4.52 (d, 1H, J_{2,3} = 0, H-2), 4.31 (dd, 1H, J_{5,6} = 6.8 Hz, H-6), 4.00 (dd, 1H, J_{4,5} = 9.6 Hz, H-4), 3.97 (dd, 1H, J_{6,6} = 10.2 Hz, H-6'), 3.69 (m, 1H, J_{5,6} = 1.7 Hz, H-5), 3.32 (m, 2H, NCH₂), 3.19 (q, 2H, NCH₂), 2.38 (s, 3H, CH₃), 1.42, 1.25 (s, 6H, C(CH₃)₂), 1.08, 1.06 (t, 6H, J = 7.0 Hz, CH₃); ¹³C NMR (CDCl₃) δ 153.9 (CO), 144.5 (C-ipso), 129.7 (2C, C-ortho, C-para), 128.0 (C-meta), 112.4 (*C*(CH₃)₂), 104.8 (C-1), 83.1 (C-2), 79.2 (C-4), 77.1 (C-3), 71.6 (C-6), 66.3 (C-5), 42.4, 41.7 (2C, NCH₂), 26.6, 26.2 (2C, C(CH₃)₂), 21.5 (CH₃), 13.8, 13.2 (2C, NCH₂/CH₃). In the second step, **6**' (2.0 g, 4.22 mmol) gave, after 1 h, 1.7 g (89%) of **15**. [α]_D²² +28° (*c* 1.2, CHCl₃).

Anal. Calcd for $C_{19}H_{34}O_6S_2N_2$ (450.62): C, 50.64; H, 7.60; S, 14.23; N, 6.22. Found: C, 50.60; H, 7.65; S, 14.26; N, 6.28.

6-Deoxy-3-*O*-(*N*,*N*-diethylthiocarbamoyl)-6-*S*-(*N*,*N*-diethyldithiocarbamoyl)-1,2-*O*-isopropylidene-α-D-glucofuranose (16). Likewise, in the first step, 7 (7.0 g, 20.9 mmol) gave 6.9 g (67%) of 3-*O*-(*N*,*N*-diethylthiocarbamoyl)-1,2-*O*-isopropylidene-6-*O*-tosyl-α-D-glucofuranose (7°). [α]_D²² -35° (*c* 1.5, CHCl₃); ¹H NMR (CDCl₃) δ 7.67 (d, 2H, $J_{0,m} = 8.3$ Hz, H-ortho), 7.20 (d, 2H, H-meta), 5.72 (d, 1H, $J_{1,2} = 3.5$ Hz, H-1), 4.80 (d, 1H, $J_{3,4} = 2.3$ Hz, H-3), 4.71 (dd, 1H, $J_{5,6} = 2.3$ Hz, H-6), 4.47 (d, 1H, $J_{2,3} = 0$, H-2), 4.29 (dd, 1H, $J_{6,6} = 11.8$ Hz, H-6'), 4.02 (dd, 1H, $J_{4,5} = 11.7$ Hz, H-4), 3.86 (m, 1H, $J_{5,6} = 2.3$ Hz, H-5), 3.61, 3.31 (q, 4H, NCH₂), 2.27 (s, 3H, CH₃), 1.26, 1.09 (s, 6H, C(CH₃)₂), 1.01, 0.99 (t, 6H, J = 7.0 Hz, CH₃); ¹³C NMR (CDCl₃) δ 190.6 (CS), 145.4 (C-ipso), 132.4 (C-para), 129.9 (C-ortho), 127.9 (C-meta), 112.2 (*C*(CH₃)₂), 104.8 (C-1), 82.6 (C-2), 81.8 (C-3), 78.5 (C-4), 71.6 (C-6), 66.8 (C-5), 47.8, 43.4 (2C, NCH₂), 26.3, 26.0 (2C, (C(CH₃)₂), 21.5 (CH₃), 13.2, 11.7 (2C, NCH₂CH₃). In the second step, 7° (5.0 g, 10.2 mmol) gave 3.4 g (72%) of 16. [α]_D²² +39° (*c* 1.0, CHCl₃).

Anal. Calcd for $C_{19}H_{34}O_5S_3N_2$ (466.69): C, 48.90; H, 7.34; S, 20.61; N, 6.00. Found: C, 49.20; H, 7.42; S, 20.45; N, 6.11.

3,6-Dideoxy-bis-3,6-S-(N,N-diethyldithiocarbamoyl)-D-glucopyranose (17). The procedure described for the preparation of 6 was applied to 2 (2.0 g, 4.1 mmol) at 60 °C for 2 h to give 0.9 g (50%) of 17. $[\alpha]_D^{22}$ -38° (c 1.5, CH₃OH).

Anal. Calcd for $C_{16}H_{30}O_4S_4N_2$ (442.68): C, 43.41; H, 6.83; S, 28.97; N, 6.33. Found: C, 43.65; H, 6.95; S, 29.30; N, 6.24.

Bis-3,6-*O*-(*N*,*N*-diethylcarbamoyl)-**D**-glucopyranose (18). Likewise, 9 (5.0 g, 12.0 mmol) gave, after 2 h, 3.1 g (69%) of 18. $\{\alpha\}_D^{27} + 52^{\circ}$ (c 1.1, CH₃OH).

Anal. Calcd for $C_{16}H_{30}O_8N_2$ (378.43): C, 50.78; H, 7.99; N, 7.40. Found: C, 50.62; H, 7.85; N, 7.35.

6-Deoxy-3-O-(N,N-diethylcarbamoyl)-6-S-(N,N-diethyldithiocarbamoyl)-D-glucopyranose (19). Likewise, 15 (5.0 g, 11.1 mmol) gave, after 2 h, 3.1 g (70%) of 19, mp 107-109 °C; $[\alpha]_D^{27}$ +47° (c 1.1, CH₃OH).

Anal. Calcd for $C_{16}H_{30}O_6S_2N_2$ (410.55): C, 46.81; H, 7.36; S, 15.62; N, 6.82. Found: C, 46.80; H, 7.45; S, 15.39; N, 6.68.

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